



Worthwhile Operational Guidelines & Suggestions

BROILER PROCESSING TIMELY INFORMATION - JUNE 2004

THE LOG SCALE (pH example)

The log (short for logarithm) is the power to which 10 is raised (or power of 10) in order to obtain the number. Example: $\log(1)=0$ (because $10^0=1$), $\log(100)=2$ (because $10^2=100$), $\log(1000)=3$ (because $10^3=1000$) and so on. The pH, which refers to the H^+

| H^+ concentration | pH | Examples |
|---------------------|----|------------------|
| 10,000,000 | 0 | Battery acid |
| 1,000,000 | 1 | Stomach acid |
| 100,000 | 2 | Lemon juice |
| 10,000 | 3 | Orange juice |
| 1,000 | 4 | Tomato juice |
| 100 | 5 | Black coffee |
| 10 | 6 | Saliva |
| 1 | 7 | Distilled water |
| 0.1 | 8 | Sea water |
| 0.01 | 9 | Baking soda |
| 0.001 | 10 | Milk of Magnesia |
| 0.0001 | 11 | Ammonia |
| 0.00001 | 12 | Soapy water |
| 0.000001 | 13 | Bleach |
| 0.0000001 | 14 | Drain cleaner |

concentration relative to distilled water (pH=7 or neutral), is expressed on a log scale that ranges from 0 (very acid) to 14 (very alkaline). It is important to remember that a one unit change in the pH scale corresponds to 10 fold change in H^+ and corresponding acidity or alkalinity. Watch that black coffee (100 times more acid than distilled water)...



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